

Particles and changes of state

Glossary

boiling point – the temperature above which a liquid becomes a gas

change of state – the process of change from one state of matter to another

condensation – the process of change from a gas into a liquid
to **condense** - verb

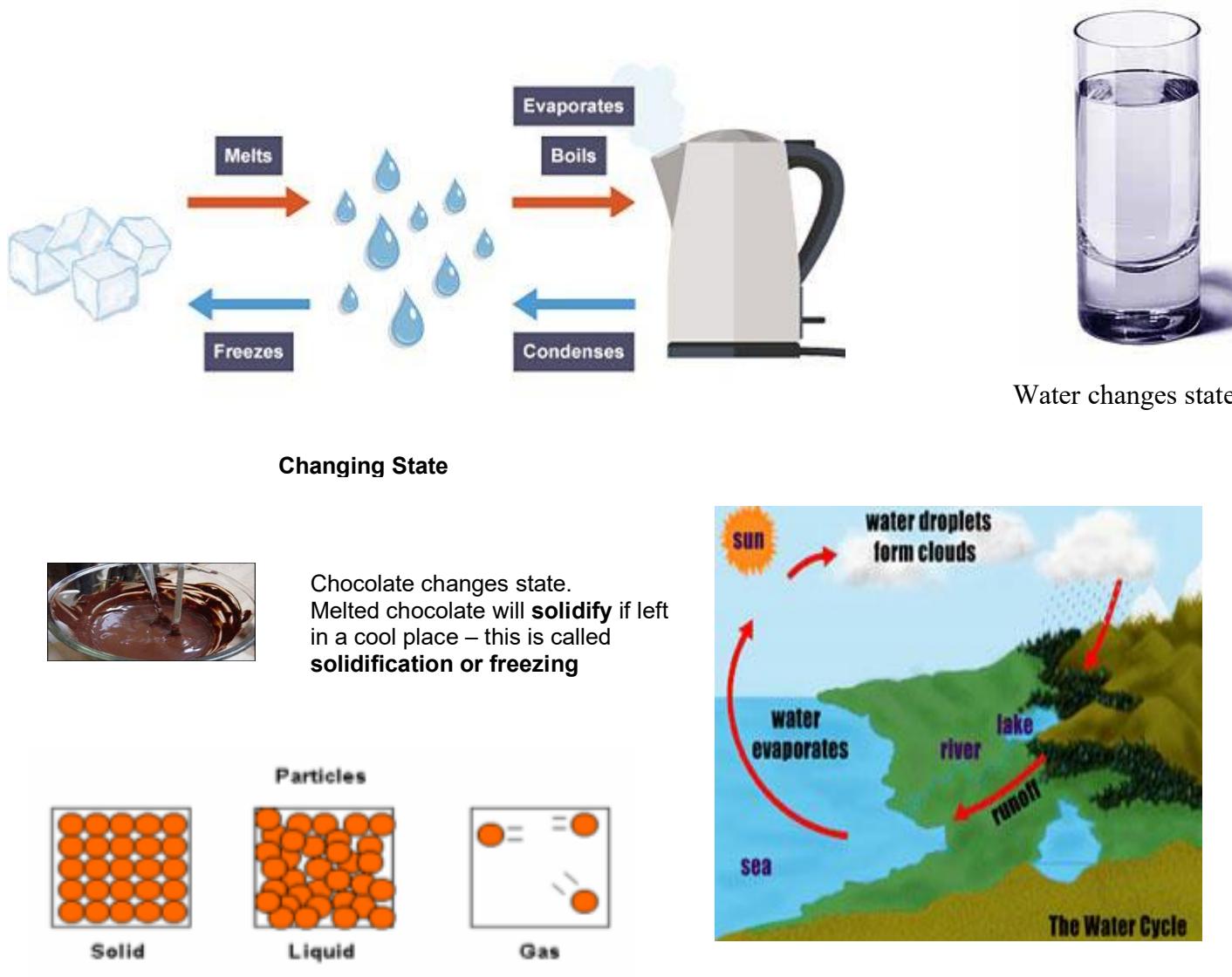
evaporation – the process of change from a liquid into a gas
to **evaporate** - verb

freezing – the process of change from a liquid into a solid
to **freeze** - verb

freezing point – the temperature below which a liquid becomes a solid – for water this is 0°C

gas – one of the three states of matter. Gases move to fill any available space. The particles in a gas are very far apart from each other and move freely

Insulate: materials used to keep the temperature the same. Materials can be ranked as poor or good. These materials make it more difficult for thermal energy to 'escape'.



liquid – one of the three states of matter. In a liquid the particles are not as close together as in the solid form. Liquids can be poured and take on the shape of the container they are placed in

melting – the process of change from a solid into a liquid to **melt** - verb

melting point - the temperature above which a solid becomes a liquid

particles-almost everything is made of these; they are very small

solid – one of the three states of matter. Solids keep their shape. The particles of a solid are very close together

solidification – the process of a liquid hardening to form a solid
to **solidify** - verb

states of matter – all material exists in three states – **solid, liquid and gas**

thermal Conductor: these materials allow heat to pass through easily. Can be ranked as poor or good.

Thermometer- a piece of equipment used to measure temperature

water cycle – the cycle of events that occur naturally in the weather systems of the Earth where water moves through its three states